SUBSTITUENT EFFECT IN THE TRANSANNULAR TC-INTERACTION OF BICYCLO(3.2.2) NONA-3, 6-DIEN-2-ONE DERIVATIVES

Shô Itô, Hitoshi Takeshita, Yoshikazu Shoji, Yasuo Toyooka and Tetsuo Nozoe

Department of Chemistry, Tohoku University

Sendai, Japan

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In the course of our study on the Diels-Alder reaction of troponoids using several dienophiles (1, 2, 3, 4, 5), we found a novel substituent effect on the transannular π -electron interaction in bicyclo-(3.2.2) nona-3, 6-dien-2-one system.

Adducts synthesized and their UV maxima are listed in TABLE 1.

The parent ring system with an electron attracting substituent at the 8a- and/or the 9a-position (compounds I-IV) invariably exhibits, as is expected (6), two maxima (FIGURE), maximum A with higher intensity at ca. 225 mµ (7) and maximum B with lower intensity at around 335 mµ. From their shift in solvents of different polarity, it is clear that the maxima A and B originate from the π - π * and n- π * transitions, respectively. Configurational changes at the 8- and 9-positions and/or introduction of various substituents at the 1-position (compounds V-XIII) cause practically no effect on maximum A, although some shifts are observed for maximum B. On the other hand, introduction of halogens (compounds XIV-XXI), methyl group (compounds XXIV, XXV) or methoxyl group (compounds XXII, XXIII) at the 3-position causes the following changes: First, the maximum A shows the bathochromic shift in the increasing order of H through CH₃, Cl and Br, and to OCH₃. However, the magnitude of the bathochromic shift is smaller than that expected from the Woodward rule (6) and the intensity of the maximum decreases in the same order, as shown in TABLE II. Secondly, a third maximum (a shoulder for Cl and CH₃ compounds), designated as maximum A', appears in the range of 260-275 mµ, the wave length and the intensity of the maximum increases in the order CH₃, Cl, Br, OCH₃ (FIGURE). The π - π *

3 0

Br

32 0 , 8p

4

TABLE 1.

Absorption Maxima of Bicyclo[3.2.2]nonen-2-one Derivatives in Methanol.

n	onen-2	2-one	Deri	vative	s in N	[ethano]	1.	7 AP	'8α F		4
Compds	•			Subst	ituent	s		6 ³ γ _{9α}	•O		
No.	No.			exo		endo		λ_{\max}^{MeOH} (mu)(log		ε)	
<mark>∆⁶serie</mark>	<u>s</u> 3	1	7	8β	9β	вα	9α	A	A '	В	Ref.
I	-	-	-	-	-	CN	-	226(3.89)	-	338(2.03)	4
II	-	-	-	. ·	-	-	CN	224(3.95)	-	340(1.95)	4
III	-	-	-	-	-	CO -	-0-00	225(3.80)	-	337(2.14)	1
IV	-	-	-	-	-	COOMe	COOMe	227(3.90)	-	335(2.30)	1
v	-	OH	-	-	-	ĊO -	-0 - CO	225(3.78)	-	523(2.07)	3
VI	-	OH	-	CO-0	0-00	-	-	225(3.80)	-	323(2.07)	3
VII	-	OMe	~	-	-	CO-	-0 - CO	224(3.76)	-	546(2.06)	3
VIII	-	OMe	-	-	-	COOMe	COOMe	226(3.80)	-	345(1.95)	3
IX	-	C1	-	CN	-	-	-	226(3.81)	-	332(1.93)	4
Х	-	OMe	-	-	-	CN	-	225(3.87)	-	345(2.03)	4
XI	-	OAc	-	-	-	CN	-	224(3.86)	-	532(2.09)	4
XII	-	OAc	-	CN	-	-	-	225(3.84)	-	352(1.79)	4
XIII	-	OAc	-	CO-0	0-00	-	-	224(3.69)	-	324(2.07)	3
XIV	Cl	-	-	-	-	co-	0-00	234(3.81)	260(3.45)*	350(2.07)	2
ΧV	Cl	-	-	-	-	COOMe	COOMe	235(3.68)	260(3.46)*	333(2.00)	2
XVI	C1	-	-	-	-	-	CN	234(3.72)	260(3.44)*	330(1.84)	4
XVII	C 1	-	-	CN	-	-	-	237(3.69)	262(3.47)*	335(1.90)	4
XVIII	Cl	-	-	-	-	CN	-	237(3.66)	262(3.49)	332(1.97)	4
XIX	Br	-	-	-	-	co -	0-00	245(3.55)	275(3.47)	337(2.13)	2
ХХ	Br	OMe	-	-	-	co -	0-00	245(3.50)	275(3.45)	340(2.15)	**
XXI	Br	OMe	-	-	-	COOMe	COOMe	248(3.50)	275(3.47)	347(1.80)	**
IIXX	0Me	-	-	-	CN	-	-	231(3.43)	276(3.48)	****	4
XXIII	OMe	-	-	-	-	-	CN	231(3.51)	276(3.58)	325(2.13)*	4
XXIV	СН3	-	-	-	-	co-	0 - C0	235(3.70)	260(3.2)*	328(1.99)	* *
XXV	сн,	-	-	-	-	COOH	COOH	236(3.78)	260(3.3)*	327(2.24)	**
XXVI	-	OMe	Br	C0-0	0-00	-	-	233(3.81)	270(2.5)*	345(2.14)	* *
Bromo-)	(-lact	one s	eries	3 ***							
XXVII	-	-						232(3.82)		320(1.65)	1
XXVIII	снз	-						240(3.81)		320(1.81)	* *
XXIX	Cl							250(3.97)		****	2
XXX	Br	-						258(3.84)		320(1.85)	2
XXXI	Br	och3						260(3.74)		****	* *

* Shoulder ** Unpublished compound prepared by Y. T.

*** Positions of bromine and lactone ring might be reversed.

**** Bands are not observed.



UV Spectra of Bicyclo [3.2.2] nona-3, 6-dien-2-one Derivatives.

ΤA	٩B	L	E	Т	Ι.	
•••		_			• •	

Substituent Effect of Absorption Spectra of Bicyclo(3.2.2) nona-3, 6-dien-2-one Derivatives.

Substituent		X=H	X=C1	X=Br	X=OMe	X=Me	Z=R
	calcd*	226	241	251	261	236	226
λ max A (mµ)	obs.	224-227	234-237	245-248	231	235-236	233
log ¢	obs.	3.69-3.95	3.66-3.81	3.50-3.55	3.43-3.51	3.70-3.78	3.81
$\wedge \lambda \frac{MeOH}{max A(ob}$	mµ) ⊳s.−calco	-2~+1 I.)	-4~-7	-3 ~ - 6	-30	-1~0	+7
λ ^{MeOH} max A' (mµ)	obs.	-	260-262**	275	276	260**	270**
log _¢ A'	obs.	-	~3.46	~3.46	~3.53	3.25	2.50
Δλ ^{MeOH} max (A'-A	(mµ)	-	25-26	27-30	45	24	37
€ _{A'} ∕ε _A		-	0.56	0.81	1.15	0.33	0.05
No. of Exam	ples	11	5	3'	2	2	1

* based on the Woodward rule

** shoulder

nature of maximum A' is indicated by measurement in solvents of different polarity. Furthermore, if one compares the frequency separation, $\Delta\lambda$ max, of maxima A and A', with the ratio $\epsilon_{A'}/\epsilon_{A}$, both are seen to increase in the same order H to OCH₃ as shown in TABLE 11. A similar effect is caused by the bromine atom on the isolated double bond (at the 7-position)(XXVI). Since these abnormalities are not observed (TABLE 1) for bromolactones (compounds XXVIII-XXXI) without the double bond between C₆ and C₇ (1,2,5), the origin of maximum A' is attributed to a transannular \mathcal{R} -interaction in the bicyclo[3.2.2]-nona-3, 6-dien-2-one system and the effect of substituent is explicable in terms of the difference in the perturbation caused by the different electron-releasing ability of the substituents, that is, the delocalization of the non-bonded electron pair in halogen or methoxyl oxygen or hyperconjugation of methyl group increases the electron density of a β -unsaturated carbonyl system and provides more overlapping with \mathcal{R} -electrons in another double bond.

As far as we are aware, no instance of such a substituent dependency in transannular interaction has been noted, and the present result may suggest that the introduction of certain substituents into barrelene (8) and homobarrelenone (9, 10) systems would favor transannular conjugation to give UV spectra considerably different from those of the parent compounds.

Reference and Footnotes

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